

Theresa I. Kückmann,  
Hans-Wolfram Lerner and  
Michael Bolte\*Institut für Anorganische Chemie, J. W. Goethe-  
Universität Frankfurt, Max-von-Laue-Strasse 7,  
60438 Frankfurt/Main, GermanyCorrespondence e-mail:  
bolte@chemie.uni-frankfurt.de

## Key indicators

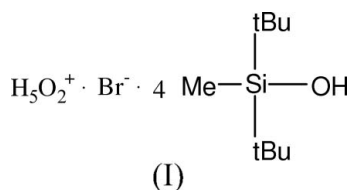
Single-crystal X-ray study  
 $T = 173$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å  
 $R$  factor = 0.049  
 $wR$  factor = 0.111  
Data-to-parameter ratio = 22.5For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Dihydronium bromide tetrakis(di-*tert*-  
butylmethylsilanol)

The asymmetric unit of the title compound,  $\text{H}_5\text{O}_2^+ \cdot \text{Br}^- \cdot 4\text{C}_9\text{H}_{22}\text{OSi}$ , is composed of four di-*tert*-butylmethylsilanol molecules, one  $\text{Br}^-$  anion, and two half  $\text{H}_5\text{O}_2^+$  ions, each of which has one disordered H atom. The silanol molecules form hydrogen bonds to the Br anion and the  $\text{H}_5\text{O}_2^+$  ions form hydrogen bonds to the hydroxyl O atoms of the silanol molecules.

Received 14 December 2006  
Accepted 18 December 2006

## Comment

Previously, we have reported the X-ray crystal structure analyses and the syntheses of supersilanol cocrystals with 18-crown-6 and water,  ${}^t\text{Bu}_3\text{SiOH} \cdot (18\text{-crown-6}) \cdot \text{H}_2\text{O}$ , and with  $\text{CF}_3\text{SO}_3\text{H}$  and water,  ${}^t\text{Bu}_3\text{SiOH} \cdot (\text{HO}_3\text{SCF}_3) \cdot 0.5\text{H}_2\text{O}$ , respectively, which were accessible from the reactions of the sodium triazenide  ${}^t\text{Bu}_3\text{Si}-\text{NNa}-\text{N}=\text{N}-\text{Si}{}^t\text{Bu}_3$  with  $\text{H}_2\text{O}$  in the presence of 18-crown-6 (Lerner *et al.*, 2002) or of  ${}^t\text{Bu}_3\text{SiO}_3\text{SCF}_3$  with  $\text{H}_2\text{O}$  (Lerner *et al.*, 2005). Surprisingly, we obtained single crystals of the title compound by hydrolysis of  ${}^t\text{Bu}_2\text{MeSiBr}$ .



A perspective view of the title compound is shown in Fig. 1. The asymmetric unit is composed of four discrete di-*tert*-butylmethyl-silanol molecules, one  $\text{Br}^-$  anion, and two half  $\text{H}_5\text{O}_2^+$  ions, each of which has one disordered H atom. The silanol molecules form hydrogen bonds to the Br anion (Table 1) and the  $\text{H}_5\text{O}_2^+$  ions form hydrogen bonds to the hydroxyl O atoms of the silanol molecules.

## Experimental

A mixture of  ${}^t\text{Bu}_2\text{MeSiBr}$  (0.23 g, 2.5 mmol),  $\text{CH}_2\text{Cl}_2$  (10 ml) and water was stored at ambient temperature. After several weeks, slow evaporation of the solvent led to the deposition of a small quantity of colourless needles.

## Crystal data

$\text{H}_5\text{O}_2^+ \cdot \text{Br}^- \cdot 4\text{C}_9\text{H}_{22}\text{OSi}$   
 $M_r = 814.36$   
Monoclinic,  $P2_1/n$   
 $a = 16.4814$  (8) Å  
 $b = 16.1107$  (5) Å  
 $c = 20.0624$  (10) Å  
 $\beta = 98.510$  (4)°  
 $V = 5268.5$  (4) Å<sup>3</sup>

$Z = 4$   
 $D_x = 1.027$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
 $\mu = 0.90$  mm<sup>-1</sup>  
 $T = 173$  (2) K  
Needle, colourless  
 $0.30 \times 0.13 \times 0.12$  mm

## Data collection

Stoe IPDS-II two-circle diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (MULABS; Spek, 2003; Blessing, 1995)  
 $T_{\min} = 0.773$ ,  $T_{\max} = 0.899$

69100 measured reflections  
 9889 independent reflections  
 8181 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.065$   
 $\theta_{\text{max}} = 25.7^\circ$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.049$   
 $wR(F^2) = 0.111$   
 $S = 1.07$   
 9889 reflections  
 440 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0494P)^2 + 4.5021P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 1.37 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.43 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O1-H1\cdots Br1$	0.69 (3)	2.57 (4)	3.256 (2)	174 (4)
$O1A-H1A\cdots Br1$	0.69 (4)	2.57 (4)	3.257 (2)	170 (4)
$O1B-H1B\cdots Br1$	0.72 (4)	2.59 (4)	3.293 (2)	163 (4)
$O1C-H1C\cdots Br1$	0.73 (4)	2.54 (4)	3.263 (2)	171 (5)
$O1W-H1WA\cdots O1C$	0.84	1.85	2.637 (3)	156
$O1W-H1WB\cdots O1^i$	0.84	1.80	2.628 (3)	169
$O1W-H1WC\cdots O1W^i$	0.84	1.60	2.437 (4)	180
$O2W-H2WA\cdots O1A$	0.84	1.84	2.634 (3)	156
$O2W-H2WB\cdots O1B$	0.84	1.85	2.639 (3)	155
$O2W-H2WC\cdots O2W^{ii}$	0.84	1.60	2.442 (4)	180

Symmetry codes: (i)  $-x + 1, -y + 1, -z + 1$ ; (ii)  $-x, -y + 1, -z + 1$ .

H atoms bonded to C were refined with fixed individual displacement parameters [ $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ ] using a riding model with  $C-H = 0.98 \text{ \AA}$ . All H atoms bonded to O could be located in a difference map. The  $\text{H}_5\text{O}_2^+$  H atoms were refined with fixed individual displacement parameters [ $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{O})$ ] using a riding model with  $O-H = 0.84 \text{ \AA}$ . The disordered H atom was refined on two equally occupied sites. The hydroxyl H atoms were freely refined.

Data collection: X-AREA (Stoe, 2001); cell refinement: X-AREA; data reduction: X-AREA; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in

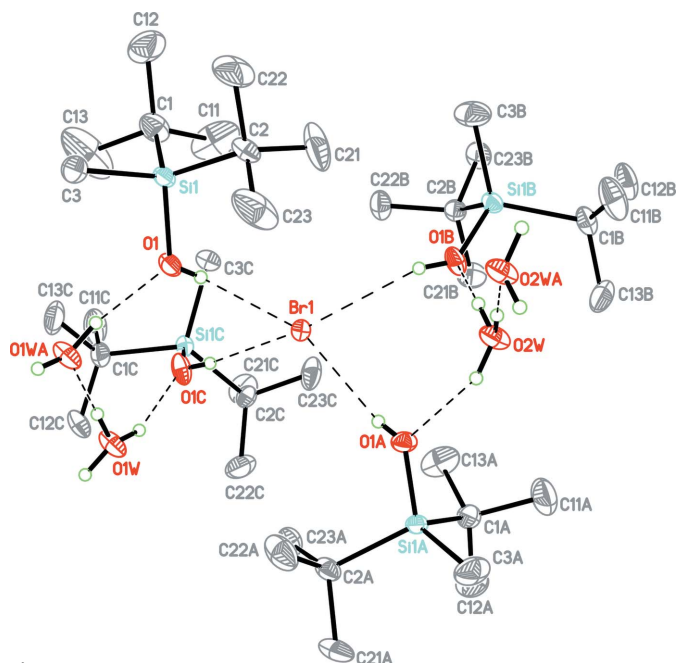


Figure 1

The asymmetric unit of the title compound, together with additional water molecules to complete the local hydrogen-bonding patterns, with displacement ellipsoids are drawn at the 30% probability level; H atoms bonded to C have been omitted for clarity. Hydrogen bonds are shown as dashed lines. Only one component of each disordered H atom of the  $\text{H}_5\text{O}_2^+$  ions is shown. Symmetry operator for O1WA is  $1 - x, 1 - y, 1 - z$  and  $-x, 1 - y, 1 - z$  for O2WA.

SHELXTL-Plus (Sheldrick, 1991); software used to prepare material for publication: SHELXL97 and PLATON.

## References

- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.  
 Lerner, H.-W., Scholz, S. & Bolte, M. (2002). *Organometallics*, **21**, 3827–3830.  
 Lerner, H.-W., Scholz, S., Wiberg, N., Polborn, K., Bolte, M. & Wagner, M. (2005). *Z. Anorg. Allg. Chem.* **76**, 1863–1870.  
 Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.  
 Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.  
 Stoe (2001). *X-AREA*. Stoe & Cie, Darmstadt, Germany.